

# **GCSE**

# **Additional Science B**

Unit B722/01: Modules B4, C4, P4 (Foundation Tier)

General Certificate of Secondary Education

Mark Scheme for June 2014

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

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1. These are the annotations, (including abbreviations), including those used in scoris, which are used when marking

Annotation	Meaning	
ВР	Blank Page – this annotation <b>must</b> be used on all blank pages within an answer booklet (structured or unstructured) and on each page of an additional object where there is no candidate response.	
<b>✓</b>	correct response	
×	incorrect response	
BOD	benefit of the doubt	
NBOD	benefit of the doubt <u>not</u> given	
ECF	error carried forward	
^	information omitted	
I	ignore	
R	reject	
CON	contradiction	

2. Abbreviations, annotations and conventions used in the detailed Mark Scheme.

/ = alternative and acceptable answers for the same marking point

(1) = separates marking points

**allow** = answers that can be accepted

not = answers which are not worthy of credit
reject = answers which are not worthy of credit

**ignore** = statements which are irrelevant

() = words which are not essential to gain credit

= underlined words must be present in answer to score a mark (although not correctly spelt unless otherwise stated)

ecf = error carried forward AW = alternative wording ora = or reverse argument

Question	Answer		Marks	Guidance
1 a	any two from:		2	
	enters through leaves (1)			
	stomata (1)			ignore pores
	by diffusion (1)			
b	herbicide (1)		1	mark answer on line first
			•	<b>allow</b> answer ringed, underlined or ticked on diagram if no
				answer on the answer line
С	crop rotation (1)		1	more than one tick scores 0
d			2	if more than 2 ticks deduct a m
	soya beans	<b>√</b> (1)		ark for each extra tick
	The chemical used on barley plants reduces the size of the sugar beet crop			
	Sugar beet grows better after soya beans than after barley			
	The growth of sugar beet is affected by both soya beans and the chemical used	<b>√</b> (1)		
	Sugar beet grows just as well after soya beans as after barley			
е	any two from:		2	
	idea that the results can be checked (1)			
	idea that the results can be compared (1)			
	to make other scientists aware of the work that habeen done (1)	as		ignore other people or public
	allows other scientists to collect further evidence other scientists can develop the work (1)	/ so		
	idea of recognition for work / improve career / become famous (1)			ignore financial remuneration
	Total		8	

Question		Ansv	ver	Marks	Guidance
2 a	air contains oxygen /			1	allow idea that oxygen gets into the compost
	oxygen is needed fo	r decay	1		
	oxygen is needed fo	r microl	oes or bacteria or fungi		
	(1)				
b	any two from:			2	
	(dissolved) in water /	in solu	tion (1)		
	by root (hairs) (1)				
	(from) the soil (1)				allow higher level answers eg active transport
c i				2	ignore references to temperature assume first reference is A if not stated
	decay in A was (at a)	faster	(rate) / ora (1)		ignore just A decays more
	idea that A finished q	ıuicker /	' ora (1)		allow less time to decompose (1)
ii				1	
	grass clippings	Α			
	sawdust	С			
	straw	В			
			(1)		
	Total			6	

Question	Answer	Marks	Guidance
3 a	[Level 3] Answer includes details of how to set up a transect line and collect the data AND contains two correct conclusions from the data. Quality of written communication does not impede communication of the science at this level.  (5 – 6 marks) [Level 2] Answer includes some description of how use a quadrat AND one correct conclusion. Quality of written communication partly impedes communication of the science at this level.  (3 – 4 marks)  [Level 1] Answer includes some reference to how the experiment is conducted OR there is an attempt at a conclusion. Quality of written communication impedes communication of the science at this level.  (1 – 2 marks)	6	This question is targeted from grades G to C  Indicative scientific points may include:  Method:  • method of sampling using quadrats to count individuals or percentage cover  • use of a key to identify organisms  • idea that a tape measure is used to set up a transect line  • quadrats used at set distances along the line  Conclusion:  • an appreciation that  • the organisms are growing at particular areas of the shore  • some organisms are more abundant than others  • some live over wider ranges  • allow references to individual species e.g.  Lower shore sawwrack, coral weed  Mid shore limpets, barnacles, coral weed bladderwrack  Upper shore bladderwrack  Use the L1, L2, L3 annotations in scoris; do not use ticks.
	[Level 0] Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)		
	Total	6	

Question	Answer	Marks	Guidance
4 a	(close to the) chloroplasts (1)	1	ignore chlorophyll
b	oxygen is given off / oxygen is a (waste) product / oxygen is produced / oxygen is made / (1)	3	allow idea that most oxygen / higher concentration of oxygen is in this area (1) ignore references to glucose
	bacteria need oxygen (so move towards it) (1)		allow bacteria attracted to the oxygen (1) but bacteria need oxygen to respire (2) ignore references to breathing
	photosynthesis is taking place in the chloroplasts / chlorophyll (1)		
С	palisade (mesophyll) (1)	1	mark answer on line first allow correct answer indicated on list if answer line is blank
	Total	5	

Question	Answer	Marks	Guidance
5 a	aluminium / gallium / indium / thallium / A $l$ / Ga / In / T $l$ (1)	1	symbols must be correct if given
b i	neutron (1)	1	
ii	<u>+5</u> (1)	1	ignore positive or + on its own
iii	neutral (1)	1	allow correct answer indicated on list if answer line is blank allow 0
	Total	4	

Question	Answer	Marks	Guidance
6	Level 3  Applies understanding of the relationship between property and use of a material to give two properties needed by the metal AND identifies metal C with two reasons Quality of written communication does not impede communication of the science at this level (5 – 6 marks)    Level 2  Applies understanding of the relationship between property and use of a material to give two properties needed by the metal OR identifies metal C with a reason Quality of written communication partly impedes communication of the science at this level (3 – 4 marks)    Level 1  Applies understanding of the relationship between property and use of a material to give one property needed by the metal OR identifies C as a suitable metal (without a reason) Quality of written communication impedes communication of the science at this level (1 – 2 marks)    Level 0  Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)	<b>7</b>	This question is targeted at grades up to C.  Indicative scientific points at all could include:  • Metal must be strong  • Metal must not corrode or rust  • Metal must be malleable  • Metal must not be brittle  Reasons:  Metal C  • idea that has highest relative strength / is very strong  • idea that corrodes slowest / is slow to corrode ignore idea that metal C is the least corrosive  ignore other properties  Use the L1, L2, L3 annotations in scoris; do not use ticks.
	1000		

Question	Answer	Marks	Guidance
7 a	O <sub>2</sub> contains two (oxygen) atoms (1)	2	ignore does not have a charge
	O <sup>2-</sup> has a charge / has a negative charge / has a 2-(1)		allow is negative / has gained electrons (1)
b	sodium atom loses an electron (1)	2	allow sodium atoms lose electrons
	to form a stable electronic structure / to get a full outer shell / to get a noble gas electronic structure / to get the electronic structure of neon / to get a stable outer octet (1)		allow more protons than electrons
С	Na <sub>2</sub> O / ONa <sub>2</sub> (1)	1	<b>allow</b> $(Na^+)_2O^{2^-}$ <b>allow</b> answer on right hand side of equation (the equation does not need to be balanced) e.g. $Na + O_2 \rightarrow Na_2O$ (1) <b>not</b> $Na2O / Na^2O$
	Total	5	

Qu	estion	Answer	Marks	Guidance
8	а	lakes / reservoirs / rivers (1)	1	allow wells / canals / streams (1) ignore oceans
	b	water wastage through leakage = 1500 (1)	2	allow 1 500 000 000
		percentage wastage = 16.7 (%) (1)		allow ecf from wrong value of water leakage allow 16.6 (%) / 17 (%) (1) not 16 (%)
	С	lead (1) from (old) lead pipes (1)	2	allow sodium fluoride, hormones etc.
		OR pesticides / nitrates / fertilisers (1) from run-off from farms (1)		allow herbicides (1) allow washed into rivers (1)
		OR		
		chlorinated compounds (1) by reaction of chlorine with impurities in water (1)		ignore chlorine or any other chemicals added in purification process
				ignore bacteria
		Total	5	

Question	Answer	Marks	Guidance
9 a	any three from:	3	maximum <b>two</b> marks for chlorine
	chlorine is used to sterilise water or purify water (1)		allow to make water safe to drink
	chlorine is used to <b>kill</b> microbes or bacteria (1)		ignore germs ignore cleaning swimming pools
	chlorine is used to make bleach / disinfectants (1)		ignore dicarming owniming pools
	chlorine is used to make pesticides (1)		
	chlorine is used to make hydrochloric acid (1)		
	chlorine is used to make plastics / used to make PVC (1)		
	iodine is used to sterilise wounds / correct medical uses (1)		
			allow iodine is used as an indicator to test for starch (1)
b	chlorine + calcium iodide → iodine + calcium chloride (1)	1	<b>allow</b> correct formulae / mix of words and formulae e.g. $Cl_2 + CaI_2 \rightarrow CaCl_2 + I_2$ <b>allow</b> reactants in either order and / or products in either order <b>not</b> calcium iodine or calcium chlorine
С	fluorine / F / F <sub>2</sub> (1)	1	allow correct answer indicated on list if answer line is blank
	Total	5	

Question	Answer	Marks	Guidance
10 a i	B (1)	1	<b>allow</b> correct answer indicated on list or diagram if answer line is blank
ii	4 Ω scores (3) 4 (2)	3	mark the unit <b>independently</b> eg. ohms / $\Omega$ scores (1) eg 100 Ohms / 100 $\Omega$ scores (1)
	but if the answer is incorrect or incomplete then: $\frac{6}{1.5}$ scores (1) $\frac{6}{1.5}$ ohms / $\Omega$ (1)		
b	any 2 from the ideas that:	2	
	current in E is higher than F (1) both bulbs in F have same current (1)		<b>allow</b> for full credit, higher level answer in terms of current eg Current in E is twice that in F (2)
	E and F have a different resistance (1)		<b>allow</b> for full credit, higher level answer in terms of resistance eg E has lower resistance / F has higher resistance (2)
			E has a higher resistance scores 1 for being different resistance
			if no other mark scored allow 1 for idea that voltage across bulb in E is greater than across each bulb in F
c i	C (1)	1	allow correct answer indicated on list if answer line is blank
ii	idea of safety (1)	1	allow higher level answers eg fuse blows if current is too large (1) ignore to stop the appliance becoming live / to stop you getting electrocuted if answer includes safety ignore explanations eg for safety to stop the appliance becoming live scores 1
	Total	8	

Question	Answer	Marks	Guidance
11 a	any two from:	2	OR allow reverse arguments for real grass:
	(artificial grass) is an insulator (1)		idea that real grass (soil) is a conductor (1)
	(Sophie) becomes charged or static (electricity) or charge is produced on artificial grass (1)		idea that charge does not accumulate / charges flow to earth (continuously) (1)
	(Sophie gets a shock) when earthed (through the post) / (Sophie gets a shock) when charge flows to ground (1)		
b	any two from:	2	
	rod needs to be an insulator (1)		if no other mark awarded allow nylon or polythene rods become charged
	cloth needs to be an insulator (1)		or rods that pick up the paper become charged (1)  OR  copper or glass or aluminium rods do not become charged (1)
	only insulators can become charged / only insulators pick up paper (1)		allow metals do not become charged
	metals do not work / are conductors / AW (1)		
	BUT both need to be insulators (2)		
С	paint or crop spraying / starting heart / defibrillator (1)	1	allow electrostatic cloths / dusters / photocopiers / printers (1)
	Total	5	

Question	Answer	Marks	Guidance
12	[Level 3] Gives a detailed description of the method AND chooses tracer E giving a correct justification. Quality of written communication does not impede communication of the science at this level (5 – 6 marks)	6	This question is targeted up to grade C  Indicative scientific points at level 3 for choice of tracer is: gamma E emitter chosen for its appropriate (long enough to detect) half-life AND penetrates soil / pipe
	[Level 2] Gives a simple or partial description of the method AND chooses tracer D or E with a supporting reason. Quality of written communication partly impedes communication of the science at this level  (3 – 4 marks)  [Level 1] Gives a simple or partial description of the		Indicative scientific points at level 2 / level 1 for choice of tracer is  E emitter chosen for its appropriate (long enough to detect) half-life  OR  D / E emitter chosen for its appropriate (short enough not to cause harm) half life  OR  D / E / gamma source chosen for soil penetration
	Gives a simple or partial description of the method involved OR chooses tracer D or E with a supporting reason Quality of written communication impedes communication of the science at this level (1 – 2 marks)  [Level 0] Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)		<ul> <li>Description at all levels may include: <ul> <li>uses a detector</li> <li>measures radiation on surface along the pipe</li> <li>blockage is where count rate changes / blockage shows a larger reading / blockage followed by a reduced reading</li> </ul> </li> <li>Use the L1, L2, L3 annotations in Scoris; do not use ticks.</li> </ul>
	Total	6	

Ques	tion	Answer	Marks	Guidance
13 a	1	fission is <b>nuclei</b> splitting (1)	2	ignore atoms splitting
		fusion is <b>nuclei</b> joining (1)		ignore atoms joining
b	)	idea that no-one has been able to repeat the experiment successfully / no-one has got the same result (1)	1	allow there were no (independent) witnesses to the cold fusion experiment / AW (1) allow no-one has been able to get cold fusion to work (1) ignore just the idea that no-one has repeated the experiment
С	; i	activity decreases with time / AW (1)  need to ensure the patient gets the correct dose / AW (1)	2	allow gets weaker with time / AW (1) ignore becomes less ionising allow takes longer to receive the correct dose / AW (1)
	ii	gamma (1)	1	accept alpha (for localised treatment) (1)
		Total	6	

Question	Answer		Marks	Guidance
14 a	radon gas, rocks, cosmic radiation (1)		2	allow in the air for radon allow from space for cosmic radiation
	85.5 (1)			ignore 86 if one of the sources is left out allow ecf if wrong source no ecf
b i	total radiation = 4120 (2)		2	<b>allow</b> one error in any figure or in addition for 1 mark correct values are (2410)+(260)+50+900+400+100
ii	any two from:			allow ecf from (b)(i)
	idea that he is receiving higher than the average dose / higher than 3000 (1)			allow because he is only just above the average (1)
	but he receives lower than the limit put on workers / lower than 20, 000 (1) he receives (far) lower than the level shown to cause cancer /			
	lower than 50,000 (1)			
c i	background reading needs to be taken off (1)	ckground reading needs to be taken off (1)		other radiation is coming from the surroundings / AW (1)
ii	40 (1)			if 40 is incorrect, allow ecf for value for radiation coming from the worktop per day x 365
	14 600 (1)			
iii	Choose a worktop made of blackstone rather than bluestone.	<b>√</b>	1	more than 2 ticks scores 0
	Put an extractor fan in her kitchen.	<b>✓</b>		
	Choose a worktop made of greystone rather than blackstone.			
	Always use gloves when she is preparing food.			
	Wear an apron in the kitchen.			
	Total		10	

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